

Answer Key To Chemical Equations

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 This example illustrates the fundamental aspects of any chemical equation: 1. The substances undergoing reaction are called reactants, and their formulas are placed on the left side of the equation. 2. The substances generated by the reaction are called products, and their formulas are placed on the right sight of the equation.

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The chemical equation has the products on the right side, while the reactants are written on the left side. Both of them are separated by an arrow. For instance, 2H2 + O2 -> 2H2O denotes that there are four atoms of hydrogen and 2 atoms of oxygen on both sides of the equation. The amount of reactants must be equal to the amount of products.

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Chemical Equations and Reactions SHORT ANSWER Answer the following questions in the space provided. 1. Match the symbol on the left with its appropriate description on the right. (aq) (a) (b) (c) (d) (f) (f) A precipitate forms. A gas forms. A reversible reaction occurs. Heat is applied to the reactants. A chemical is dissolved in water.

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Balancing Chemical EquationsFor your answer, type in the numbers that go in the blanks, with a comma and a space between each number (Ex. "2, 1, 3"). Terms in this set (44) 1, 1, 1. __CaO + __H2O ? __Ca(OH)?.