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Neutralization Goals : To
gain an understanding of :

1. The properties of acids and bases
2. pH and pOH calculations
3. Definitions of acids and bases
- 4.

Neutralization reactions

NOTES: Properties of acids :

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- Form hydrogen ions (H^+ or protons) in solution

CHEMISTRY NOTES - CHAPTERS
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...

Chapter 20 acids and bases.
the particle formed when a

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base gains a hydrogen ion:
 NH_4^+ is the conjugate acid
of the base NH_3 *20.3. two
substances that are related
by the loss or gain of a
single hydrogen ion.

Chapter 20 Acid And Bases

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Describe the effects of phenolphthalein on both acids and bases. 14. What is neutralization? 15. What is a salt? 16. What are the two products of a neutralization reaction? 17. Copy the chart

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on page 244. 18. What happens when an acid and a base react in water? 19. What are the proton-donor and proton-acceptor definitions for acids and bases? 20.

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