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Answer: (i) n = 3 Thus element belong to 3rd period, p-block element. Since the valence shell contains = 6 electrons, group No = 10 + 6 = 16 configuration =1s 2 2s 2 2p 6 3s 2 3p 4 element name is sulphur. (ii) n = 4 Means element belongs to 4th period belongs to group 4 as in the valence shell (2 + 2) = 4 electrons.

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