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(1000 g) of corks contains (1000 g

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1. Why standard entropy of an elementary substance is not zero whereas the standard enthalpy of formation is taken as zero? Answer: A substance has a perfectly ordered arrangement only at absolute zero. Hence, entropy is zero only at ...

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the reaction we monitor the mass of Ca^{2+} remaining in solution and the mass of CaCO_3 that precipitates, the result looks something like Figure 6.1. At the start of the reaction the mass of Ca^{2+} decreases and the mass of CaCO_3 increases. Eventually the reaction reaches a point after which there

Chapter 6

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