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to form a single new substance. As shown in Figure 11.5, magnesium metal and

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Unit 2 - Chemical Reactions

Balance the following reactions and indicate which of the five types of chemical reaction are being represented: 1) _____ NaBr + _____ Ca(OH) 2 _____ CaBr 2 + _____ NaOH

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bonds. 8. It is a reaction that releases energy. 9. It is a reaction that absorbs energy. 10. a. Yes.
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REACTIONS KEY For each of the following reactants, predict whether a precipitation reaction will take place between them If there is no reaction, write “ no reaction ” If there is a reaction, write the complete, complete ionic, and net ionic equations that describe the reaction

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