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Equilibrium 2--Calculating Equilibrium ~~The Equilibrium Constant~~
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Chapter 15 – Chemical Equilibrium: Part 9 of 12 Chemical
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Chapter 15 – Chemical Equilibrium: Part 4 of 12 **Modern**

Chemistry Chapter 15 Test

Modern Chemistry 136 Chapter Test Name Class Date Chapter Test
A, continued ____ 14. A pH meter measures the pH of a solution
by measuring the a. amount of charge on the hydronium ion. b.
color of the solution. c. concentration of the solution. d. voltage
difference between two electrodes in the meter's probe. ____ 15.

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When you conduct an ...

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Modern Chemistry Chapter 15. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. mhardie11. Terms in this set (10) Self-ionization of water. two water molecules produce a hydroniumion and a hydroxide ion by transfer of a proton. pH.

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Holt McDougal Modern Chemistry Chapter 15: Acid-Base ...

Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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Chapter 15 Acids Bases Test Modern Chemistry

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Modern Chemistry Chapter 15 Test Review 25 multiple choice definition of self-ionization $[H_3O^+] [OH^-] = 1 \times 10^{-14}$ definition and formula for calculating $pH = -\log[H_3O^+]$ $pH > 7$ is a base; $pH < 7$ is an acid; $pH = 7$ is neutral pH range is normally 0 to 14 calculate pH from $[H_3O^+]$ calculate $[H_3O^+]$ from pH definitions of

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indicators, transition interval, titration, neutralization, endpoint, standard solution acid-base titration monitors pH titration problems using $V_a M_a e_{q_a} = V_b M_b e_{q_b}$.

Modern Chemistry Chapter 15 Acid-Base Titration & pH

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Bases, and Solutions (continued) 28. Dip red litmus paper into each solution and watch for color changes. Then do the same with blue litmus paper. Any solution that turns red litmus blue is basic. Any solution that turns blue litmus red is acidic.

Chapter 15 Acids Bases Test Modern Chemistry

Modern Chemistry 105 Chapter Test Name _____ Class _____ Date _____ Chapter Test A, continued Use this figure to answer questions 7 and 8. _____ 7. A solution containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C to 90°C . According to information in the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in ...

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Test Answers

A, continued _____21. Metal X replaces the ions of metal Y from solution, but it cannot replace the ions of metal Z from solution. The order these metals should have in the activity series (from top to bottom) is a. x, y, z b. z, x, y c. y, x, z d. z, y, x _____22.

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Modern Chemistry 101 Chapter Test Name _____ Class _____ Date _____ Chapter Test B, continued 18. The pressure exerted by each gas in a mixture is called the _____ of that gas. 19. If the temperature and number of moles of a gas remain constant but the volume increases, the pressure of the gas will _____. 20. The lowest possible temperature, corresponding to zero on the ...

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