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## CHEMISTRY

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Answers  
mass relationship between the reactants and products of a chemical reaction. You have been studying composition stoichiometry in the previous chapters. Our study of chemical reactions introduces mass relationships ...

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*STOICHIOMETRY deals ...*

modern chemistry stoichiometry chapter 9 vocab. reaction stoichiometry. composition stoichiometry. what does a coefficient tell you. mole ratio. chemistry that deals with mass relationships between reactants... deals with mass relationships of elements in compounds. amount of atoms or moles in a substance.

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CHAPTER 9 REVIEW. Stoichiometry. SECTION 1. SHORT ANSWER Answer the following questions in the space provided. 1. bThe coefficients in a chemical equation represent the. (a) masses in grams of all reactants and products. (b) relative number of moles of reactants and products.

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REVIEW Stoichiometry SECTION 3 PROBLEMS Write  
the answer on the line to the left. Show all  
your work in the space provided. 1. \_\_\_\_\_ The  
actual yield of a reaction is 22 g and the  
theoretical yield is 25 g. Calculate the  
percentage yield. 2. 6.0 mol

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the answer on the line to the left. Show all  
your work in the space provided. 1. \_\_\_\_\_ The  
actual yield of a reaction is 22 g and the

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theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N<sub>2</sub> are mixed with 12.0 mol of H

CHAPTER 9 REVIEW ...

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stoichiometry: using a balanced chemical  
equation to calculate the number of grams,  
moles, or particles of reactants/products  
involved in a...

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