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Topic 12: Nuclear Chemistry Pg 111 \u2013 116 Worksheet 52: Definition and facts of nuclear chemistry Worksheet 53: Nuclear transmutations and equations -life calculations and Reference Table N Table of Contents .

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Topic 12 Nuclear Chemistry \u2013Chemistry \u2013Nuclear Chemistry \u2013Obj: SWBAT explain what particles are responsible for radioactivity and how the radioisotopes react. \u2013Do Now: What do you think are some advantages of using nuclear chemistry? Stability of Nuclei

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nuclear reaction. Success Criteria \u2013 Determine the amount of radioactive material remaining after some period of time. \u2013 Correctly use the relationship between energy and mass to calculate nuclear binding energies and the energy released in nuclear reactions. Resources Chemistry Matter and Change pp. 804-834 Chemistry the Central Science p ...

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Book Topic 12 Nuclear Chemistry Workbook Answers Chemistry 217 214 82 214 83 electron during the conversion of a neutron to a proton When a nucleus emits a beta particle, which has a charge of  $1^-$ , the charge on the nucleus increases by one, which

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A balanced nuclear reaction equation indicates that there is a rearrangement during a nuclear reaction, but of subatomic particles rather than atoms. Nuclear reactions also follow conservation laws, and they are balanced in two ways: The sum of the mass numbers of the reactants equals the sum of the mass numbers of the products.

### 21.2 Nuclear Equations □ Chemistry

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-Review Book: Topic 12: Nuclear Chemistry (pages 215-228)-Text Book: Ch 28: Nuclear Chemistry- pages 840-861-Check the video (link below) about types of transmutation and types of radiation-Review the unit vocabulary and skills (green sheet or see link below). Be sure you're not just reading the list.

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